

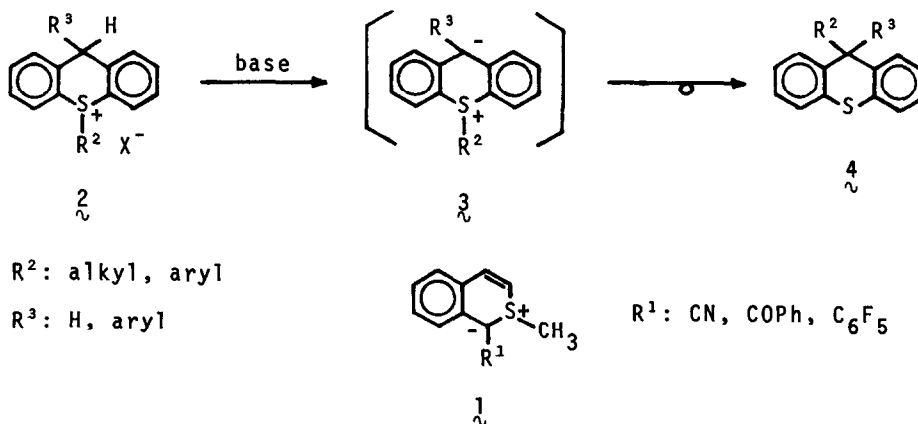
SYNTHESIS AND PROPERTIES OF THE FIRST CRYSTALLINE 1,4-YLIDE,
9-CYANO-10-METHYL-10-THIAANTHRACENE¹⁾

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Recently, much interest has been focussed on the chemistry of thiabenzenes, in particular, the preparation and properties of the stable thiabenzene derivatives. 1-Cyano-,^{1,2)} benzoyl-,^{1,2)} and pentafluorophenyl-2-methyl-2-thiaphthalenes³⁾ (1) having electron-attracting groups at 1-position, were isolated as stable compounds. However, cyclic sulfur 1,4-ylides, namely, 9,10-disubstituted 10-thiaanthracenes (3) formed by the treatment of 9,10-disubstituted thioxanthenium salts (2) with bases are quite unstable compounds which undergo thermal Stevens-type 1,4-rearrangement reaction to yield the corresponding 9,9-disubstituted thioxanthenes (4).^{1,3,4)} Hence, no report has been described on the stable thiaanthracene derivatives so far.

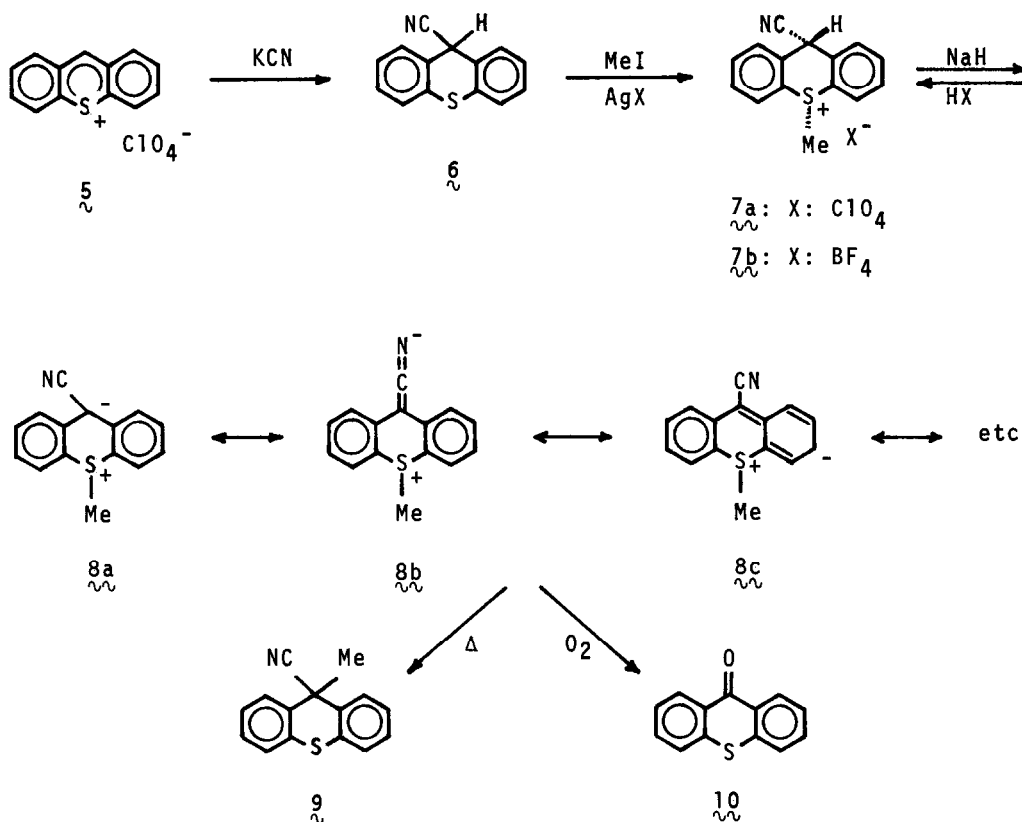


Scheme I

In this communication, we wish to report the first synthesis of a crystalline cyclic sulfur 1,4-ylide, 9-cyano-10-methyl-10-thiaanthracene (**8**) by the reaction of 9-cyano-10-methylthioxanthenium salts (**7**) with NaH.^{1,5)}

Treatment of thioxanthylium perchlorate (**5**)⁶⁾ with KCN in H₂O-CH₂Cl₂ afforded 9-cyanothioxanthene (**6**)⁷⁾ in 95 % yield, which was methylated with MeI in the presence of AgClO₄ or AgBF₄ to give 9-cyano-10-methylthioxanthenium perchlorate (**7a**)⁸⁾ with colorless prisms (from acetone), mp 181-183° (decomp.) in 54 % yield: NMR (CF₃CO₂H) δ 3.48 (3H, s, Me), 6.25 (1H, s, C₉-H), and 7.73-8.46 (8H, m, ArH); IR (KBr) ν max cm⁻¹ 2240 (CN) and 1147-1082 (ClO₄⁻), and tetrafluoroborate (**7b**)⁸⁾ with colorless scales (from acetone-ether), mp 186-189° (decomp.) in 64 % yield: NMR (CF₃CO₂H) δ 3.46 (3H, s, Me), 6.22 (1H, br. s, C₉-H), and 7.76-8.49 (8H, m, ArH); IR (KBr) ν max cm⁻¹ 2240 (CN) and 1062-1033 (BF₄⁻), respectively. Deprotonation of both **7a** and **7b** using NaH in THF under nitrogen atmosphere yielded 9-cyano-10-methyl-10-thiaanthracene (**8**), which is the first example of a stable and crystalline 1,4-ylide in thiaanthracene system, with orange-yellow needles (from CH₂Cl₂-ether), mp 115-116° in 94 % yield: NMR (CDCl₃) δ 2.22 (3H, s, Me), 6.73-7.06 (2H, m, C₃ and 6-H), and 7.19-7.77 (6H, m, ArH); IR (KBr) ν max cm⁻¹ 2170 (CN); Field Desorption(FD) MS (m/e) 237 (M⁺) which is supported as monomeric. The absorption band of cyano group in the IR spectrum of **8** is stronger and shifts to lower wave number compared with that of ordinary cyano group, indicating the delocalization of the carbanion electron of **8** through the cyano group.

Thiaanthracene (**8**) was acidified with HClO₄ or HBF₄ to afford the starting thioxanthenium salt (**7a**) or (**7b**), respectively. On standing in THF at 50° for 5 hr under nitrogen atmosphere, **8** gave thermal 1,4-rearrangement product, 9-cyano-9-methylthioxanthene (**9**)¹¹⁾ with colorless prisms (from EtOH), mp 68-69° in 82 % yield: NMR (CDCl₃) δ 1.78 (3H, s, Me), 7.19-7.60 (6H, m, Ar₂₋₇-H), and 7.78-8.11 (2H, m, Ar₁ and 8-H); IR (KBr) ν max cm⁻¹ 2250 (CN), which was also synthesized by the reaction between 9-methylthioxanthylium perchlorate⁶⁾ and KCN-H₂O in CH₂Cl₂. When **8** was stirred in THF till it became colorless, thioxanthone (**10**)¹⁰⁾ was afforded by the autoxidation as colorless needles (from benzene), mp 218-220°, (lit⁹⁾ mp 219°) in 88 % yield.



Scheme II

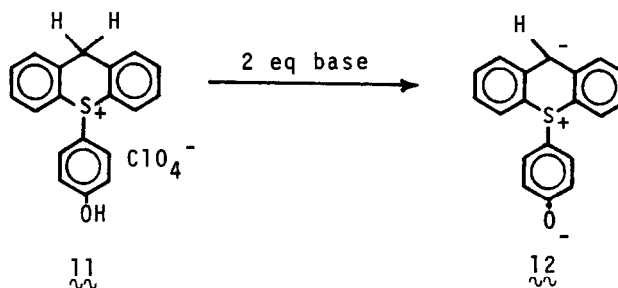
The very interesting reaction of **8** with some electrophiles will be also reported in the quite near future.¹⁰⁾

REFERENCES AND FOOTNOTES

- 1) A part of this work was presented at a) the 5th International Congress of Heterocyclic Chemistry, Ljubliana, Yugoslavia, July, 1975, Abstracts of Papers p. 299; b) the 8th Congress of Heterocyclic Chemistry, Kyoto, Japan, Oct., 1975, Abstracts of Papers p. 119 by M. Hori, T. Kataoka, H. Shimizu, S. Ohno, K. Narita, and H. Aoki.
- 2) M. Hori, T. Kataoka, H. Shimizu, K. Narita, S. Ohno, and H. Aoki, Chemistry

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- 3) a) B. E. Maryanoff, J. Stackhouse, G. H. Senkler, Jr., and K. Mislow, J. Amer. Chem. Soc., 97, 2718 (1975); b) J. Stackhouse, B. E. Maryanoff, G. H. Senkler, Jr., and K. Mislow, ibid., 96, 5650 (1974).
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- 5) Mislow and his co-workers generated 12 with remarkable stability in solution, by the treatment of 11 with 2 eq of dimethylsodium in DMSO, but they could not isolate it.^{4c)}



- 6) C. C. Price, M. Hori, T. Parasaran, and M. Polk, J. Amer. Chem. Soc., 85, 2278 (1963).
- 7) J. F. Muren, J. Med. Chem., 13, 140 (1970).
- 8) In the Scheme II, dotted lines drawn in the figure of compound 7 indicate pseudo-equatorial bonds. Configuration of 7a is cis form, and that of 7b is cis form in the state of crystal, but in $\text{CF}_3\text{CO}_2\text{H}$ solution, a part of cis form changes to trans form (cis/trans = 1.55). Trans 7b: NMR ($\text{CF}_3\text{CO}_2\text{H}$) δ 3.67 (3H, s, Me) and 6.07 (1H, s, C₉-H).
- 9) M. Gomberg and W. Minnis, J. Amer. Chem. Soc., 43, 1940 (1921).
- 10) M. Hori, T. Kataoka, H. Shimizu, and S. Ohno, Heterocycles, Vol. 7, special issue dedicated to Prof. R. B. Woodward on the occasion of his 60th birthday, in press.
- 11) Satisfactory analytical data were obtained for all new compounds.